

Exhibit B

# **TRANSPORT PHENOMENA**

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## Preface

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This book is intended to present the subjects of transport (heat conduction, port (diffusion). In this t phenomena are occurring & said about the molecular continuum approach is of dents, although it should be for complete mastery of the

Because of the current de emphasis on understanding use of empiricism, we feel tl kind. Obviously the subje across traditional departme subject of transport phenom mechanics, and electromag ences." Knowledge of the transport has certainly bec gineering analysis. In add interest to some who are meteorology, and biology.

THERMAL CONDUCTIVITY OF SOME SOLIDS AT ZERO PRESSURE<sup>a</sup>

| Substance | Thermal Conductivity<br><i>k</i><br>(cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> ) |
|-----------|---|
|           | 0.000378  |
|           | 0.000363  |
|           | 0.000247  |
|           | 0.000328  |
|           | 0.000400  |
|           | 0.000703  |
|           | 0.00143   |
|           | 0.00156   |
|           | 0.00160   |

*Physics*, Thirty-ninth Edition, Chemical  
D, pp. 2257-2259. Corrected data for  
shers.

THERMAL CONDUCTIVITIES OF SOLIDS AT SPHERICAL PRESSURE<sup>a</sup>

| Substance | Thermal Conductivity<br><i>k</i><br>(cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> ) |
|-----------|---|
|           | 0.247   |
|           | 0.290   |
|           | 0.106   |
|           | 0.119   |
|           | 0.039   |
|           | 0.037   |
|           | 0.036   |
|           | 0.0196  |
|           | 0.0261  |
|           | 0.0303  |
|           | 0.1073  |
|           | 0.0956  |
|           | 0.0846  |
|           | 0.2055  |
|           | 0.1809  |
|           | 0.1596  |
|           | 0.0617  |
|           | 0.0648  |
|           | 0.0675  |

Vol. 2, Atomic Energy Commission,  
Office, Washington, D.C. (May, 1955),

TABLE 8.1-4  
EXPERIMENTAL VALUES OF THERMAL CONDUCTIVITIES OF  
SOME SOLIDS<sup>a</sup>

| Substance             | Temperature<br><i>T</i> (° C) | Thermal Conductivity<br><i>k</i><br>(cal sec <sup>-1</sup> cm <sup>-1</sup> (° K) <sup>-1</sup> ) |
|-----------------------|-------------------------------|---|
| Aluminum              | 100                           | 0.492   |
|                       | 300                           | 0.64  |
|                       | 600                           | 1.01  |
| Cadmium               | 0                             | 0.220   |
|                       | 100                           | 0.216   |
| Copper                | 18                            | 0.918   |
|                       | 100                           | 0.908   |
|                       | 18                            | 0.112   |
| Steel                 | 100                           | 0.107   |
|                       | 100                           | 0.1528  |
| Tin                   | 0                             | 0.143   |
|                       | 100                           | —   |
| Brick (common red)    | —                             | 0.0015  |
| Concrete (stone)      | —                             | 0.0022  |
| Earth's crust (av.)   | —                             | 0.004   |
| Glass (soda)          | 200                           | 0.0017  |
| Graphite              | —                             | 0.012   |
| Sand (dry)            | —                             | 0.00093   |
| Wood (fir)            | —                             | —   |
| parallel to axis      | —                             | 0.00030   |
| perpendicular to axis | —                             | 0.00009   |

<sup>a</sup> Data taken from the *Reactor Handbook*, Vol. 2, Atomic Energy Commission, AECD-3646, U.S. Government Printing Office, Washington, D.C. (May, 1955), pp. 1766 *et seq.*

Substitution in Eq. 8.1-1 then gives

$$k = \frac{QY}{A \Delta T} = \frac{0.717 \times 0.640}{929 \times 2.00} \frac{(\text{cal sec}^{-1})(\text{cm})}{(\text{cm}^2)(\text{° K})} = 2.47 \times 10^{-4} \text{ cal sec}^{-1} \text{ cm}^{-1} (\text{° K})^{-1}$$

For  $\Delta T$  as small as this, it is usually reasonable to assume that the value of  $k$  applies at the average temperature  $(T_1 + T_0)/2$ , which in this case is 25° C. See Problems 9.F and 9.J for methods of allowing for variation of  $k$  with  $T$ .

### §8.2 TEMPERATURE AND PRESSURE DEPENDENCE OF THERMAL CONDUCTIVITY IN GASES AND LIQUIDS

The scarcity of reliable thermal conductivity data for fluids frequently makes it necessary to estimate  $k$  from other data on the given substance. We present here two correlations to aid in such estimation and to illustrate how

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